

Propane (C₃H₈) and Butane (C₄H₁₀) – Key Information

Propane (C₃H₈):

- Boiling point: -42°C
- Commonly used in LPG (liquefied petroleum gas)
- Performs well in cold environments due to its low boiling point

Butane (C₄H₁₀):

- Boiling point: -0.5°C
- Also used in LPG
- More suitable for warmer environments compared to propane

Comparison:

- Propane has a significantly lower boiling point than butane, making it more effective in colder climates
- Both gases are widely used in LPG and serve various industrial and household applications

Summary:

The document highlights differences in boiling points and environmental suitability, emphasizing how both gases contribute to the versatility of LPG systems.